

2,4,6-Trimethylacetanilide

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Key indicators

Single-crystal X-ray study
T = 293 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.045
wR factor = 0.116
Data-to-parameter ratio = 17.9For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title compound, $\text{C}_{11}\text{H}_{15}\text{NO}$, also known as *N*-(2,4,6-trimethylphenyl)acetamide, is an organic non-linear optical material. It crystallizes in the monoclinic system, in a non-centrosymmetric space group *Pn*.Received 31 July 2002
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Comment

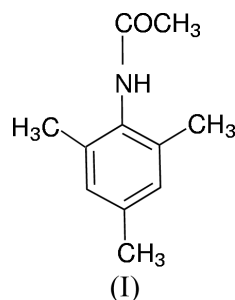
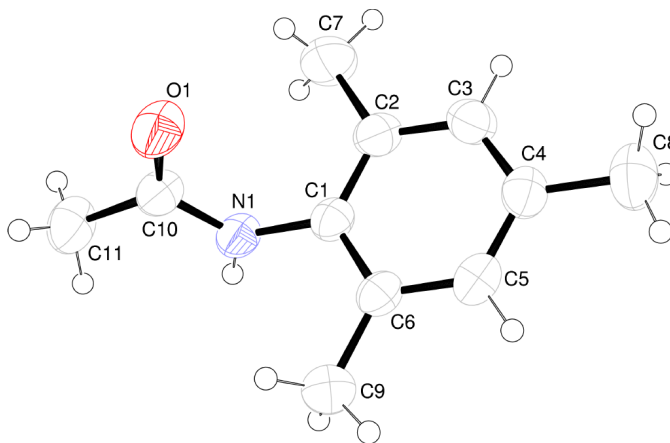
The title compound, (I), is found to be of interest as an organic non-linear optical material. The crystals show optical second harmonic generation with the fundamental beam ($\lambda = 1064 \text{ nm}$) of the Nd–YAG laser. The single-crystal structure elucidation in a non-centrosymmetric space group further reinforces this observation.Fig. 1 shows the title molecule. The torsion angle about the C1–N1 bond is $71.4(3)^\circ$, showing that the amide group is almost perpendicular to the plane of the phenyl ring (Table 1). The packing shows intermolecular (N–H...O) hydrogen bonds running along [101] (Fig. 2 and Table 2).

Figure 1

View of the molecule of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

Experimental

The title compound, (I), was prepared by the direct reaction of 2,4,6-trimethylaniline and acetic anhydride at room temperature for 10 min. Crystals of (I), suitable for single-crystal diffraction study, were grown at ambient temperature by slow evaporation of a methanol solution. The compound crystallizes as colorless prisms.

Crystal data

$C_{11}H_{15}NO$	D_m not measured
$M_r = 177.24$	Mo $K\alpha$ radiation
Monoclinic, Pn	Cell parameters from 775 reflections
$a = 8.200$ (4) Å	$\theta = 3.6$ – 26.2°
$b = 8.538$ (4) Å	$\mu = 0.07$ mm $^{-1}$
$c = 8.295$ (4) Å	$T = 293$ (2) K
$\beta = 113.626$ (6) $^\circ$	Prism, colorless
$V = 532.1$ (4) Å 3	$0.52 \times 0.44 \times 0.21$ mm
$Z = 2$	
$D_x = 1.106$ Mg m $^{-3}$	

Data collection

Bruker SMART CCD area-detector diffractometer	$R_{int} = 0.021$
φ and ω scans	$\theta_{max} = 28.0^\circ$
4076 measured reflections	$h = -9 \rightarrow 10$
2182 independent reflections	$k = -10 \rightarrow 11$
1932 reflections with $I > 2\sigma(I)$	$l = -10 \rightarrow 10$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.066P)^2 + 0.0215P]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.116$	$(\Delta/\sigma)_{max} = 0.010$
$S = 1.13$	$\Delta\rho_{max} = 0.17$ e Å $^{-3}$
2182 reflections	$\Delta\rho_{min} = -0.15$ e Å $^{-3}$
122 parameters	Absolute structure: Flack (1983),
H-atom parameters constrained	891 Friedel pairs
	Flack parameter = -0.9 (17)

Table 1

Selected geometric parameters (Å, $^\circ$).

O1–C10	1.233 (2)	N1–C10	1.350 (2)
N1–C1	1.444 (2)		
C1–N1–C10	124.57 (14)	N1–C10–C11	115.19 (15)
N1–C1–C6	118.33 (14)	O1–C10–N1	123.38 (15)
N1–C1–C2	120.71 (15)	O1–C10–C11	121.44 (16)
C10–N1–C1–C2	-71.4 (3)	C10–N1–C1–C6	110.12 (19)

Table 2

Hydrogen-bonding geometry (Å, $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N1-H1 \cdots O1^i$	0.86	2.07	2.899 (2)	162

Symmetry code: (i) $\frac{1}{2} + x, 1 - y, \frac{1}{2} + z$.

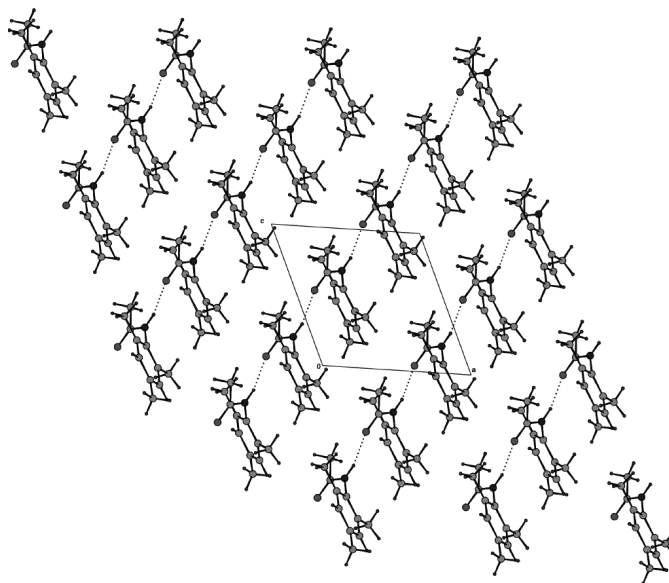


Figure 2

Packing diagram of (I), viewed down the b axis.

The absolute configuration could not be determined by the Flack (1983) test. Friedel pairs were merged.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* for Windows (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *PLATON* (Spek, 1990).

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